

Active Site Environment of Heme-Bound Amyloid β Peptide Associated with Alzheimer's Disease

Debajyoti Pramanik and Somdatta Ghosh Dey*

Department of Inorganic Chemistry, Indian Association for the Cultivation of Science, Jadavpur, Kolkata, India 700032

Received September 18, 2010; E-mail: icsgd@iaccs.res.in

Abstract: Recent reports show that there is a large increase in heme in the temporal brain of Alzheimer's disease (AD) patients, as heme, biosynthesized in brain cells, binds to amyloid β ($A\beta$), forming heme– $A\beta$ complexes. This leads to the development of symptoms that are characteristic pathological features of AD, e.g., abnormal iron homeostasis, decay of iron regulatory proteins, dysfunction in mitochondrial complex IV, oxidative stress, etc. However, the active site resulting from heme binding to $A\beta$ is not well characterized. For example, the coordinating residue, relevant second-sphere residues, and spin state of the Fe center are not known. In this study we have used wild-type and mutated $A\beta$ peptides and investigated their interaction with naturally occurring heme. Our results show that, out of several possible binding sites, His¹³ and His¹⁴ residues can both bind heme under physiological conditions, resulting in an axial high-spin active site with a trans axial water-derived ligand. Peroxidase assays of these heme–peptide complexes along with pH perturbations indicate that Arg⁵ is a key second-sphere residue that H-bonds to the trans axial ligand and is responsible for the peroxidase activity of the heme– $A\beta$ complexes. The His¹³ and Arg⁵ residues identified in this study are both absent in rodents, which do not show AD, implicating the significance of these residues as well as heme in the pathology of AD.

Introduction

Alzheimer's disease (AD) is a neurodegenerative disorder causing senile dementia with no definite known cause.¹ With the increase of general life expectancy, AD has become a severe problem for the elderly. The major symptoms of AD are long- and short-term memory loss, confusion, mood swing, and decline of sense and thinking ability.^{2–4} These symptoms generally appear at a later stage; thus, diagnosis in the early stage is difficult. AD is characterized by loss of neurons and breakdown of synaptic function in the hippocampus cortex, and this part of the neuronal cortex plays an important role in the formation of new memories.¹

The two markers of AD are the aggregation of amyloid β ($A\beta$) peptides^{1,5,6} and the generation of reactive oxygen species.^{7–9} $A\beta$ is derived by proteolysis of membrane-bound amyloid precursor protein. The long hydrophobic tail of $A\beta$ (Gly29–Val40 or Gly29–Ala42) increases its hydrophobicity and reduces its solubility. Both $A\beta(1–40)$ and $A\beta(1–42)$ are

neurotoxic in aggregated and soluble oligomeric forms which induce oxidative stress by the spontaneous generation of partially reduced oxygen species (PROS), e.g., H_2O_2 and other free radicals.^{10,11} During this process of PROS generation, $A\beta$ gets oxidized, forming proteolysis-resistant soluble dimers cross-linked via tyrosine or histidine residues of the $A\beta$ peptides.¹² The H_2O_2 generated is freely permeable across cell membranes and in the presence of reduced transition-metal ions (Fe^{2+} , Cu^+) can generate neurotoxic hydroxyl radicals.¹³ In fact, brains of patients with AD contain higher levels of Fe, Cu, and Zn.¹⁴ It has been suggested from chemical analysis of several brain samples of AD patients that the generation of H_2O_2 through the reduction of O_2 by transition metals (e.g., Cu and Fe) precedes the formation of the amyloid aggregates and hence could be responsible for the early oxidative damages observed in AD.^{15,16} This is further supported by recent developments where Cu is shown to bind $A\beta$ peptides and form well-defined active sites.^{17–19} The role and mechanism of action of the

- (1) Rauk, A. *Chem. Soc. Rev.* **2009**, *38*, 2698.
- (2) Maurer, I.; Zierz, S.; Moller, H. J. *Neurobiol. Aging* **2000**, *21*, 455.
- (3) Blass, J. P. *J. Neurosci. Res.* **2001**, *66*, 851.
- (4) Nelson, P. G. *Curr. Alzheimer Res.* **2005**, *2*, 497.
- (5) Hardy, J.; Selkoe, D. J. *Science* **2002**, *297*, 353.
- (6) Glenner, G. G.; Wong, C. W. *Biochem. Biophys. Res. Commun.* **1984**, *120*, 885.
- (7) Masters, C. L.; Simms, G.; Weinman, N. A.; Multhaup, G.; McDonald, B. L.; Beyreuther, K. *Proc. Natl. Acad. Sci. U.S.A.* **1985**, *82*, 4245.
- (8) Huang, X.; Atwood, C. S.; Hartshorn, M. A.; Multhaup, G.; Goldstein, L. E.; Scarpa, R. C.; Cuajungco, M. P.; Gray, D. N.; Lim, J.; Moir, R. D.; Tanzi, R. E.; Bush, A. I. *Biochemistry* **1999**, *38*, 7609.
- (9) Opazo, C.; Huang, X.; Cherny, R. A.; Moir, R. D.; Roher, A. E.; White, A. R.; Cappai, R.; Masters, C. L.; Tanzi, R. E.; Inestrosa, N. C.; Bush, A. I. *J. Biol. Chem.* **2002**, *277*, 40302.

- (10) Bush, A. I. *Trends Neurosci.* **2003**, *26*, 207.
- (11) Guilloreau, L.; Combalbert, S.; Sournia-Saquet, A.; Mazarguil, H.; Faller, P. *ChemBioChem* **2007**, *8*, 1317.
- (12) Smith, D. P.; Smith, D. G.; Curtin, C. C.; Boas, J. F.; Pilbrow, J. R.; Ciccosto, G. D.; Lau, T.-L.; Tew, D. J.; Perez, K.; Wade, J. D.; Bush, A. I.; Drew, S. Z.; Separovic, F.; Master, C. L.; Cappai, R.; Barnham, K. J. *J. Biol. Chem.* **2006**, *281*, 15145.
- (13) Curtin, C. C.; Ali, F.; Volitakis, I.; Cherny, R. A.; Norton, R. S.; Beyreuther, K.; Barrow, C. J.; Masters, C. L.; Bush, A. I.; Barnham, K. J. *J. Biol. Chem.* **2001**, *276*, 20466.
- (14) Bush, A. I. *Alzheimer Dis. Assoc. Disord.* **2003**, *17*, 147.
- (15) Lynch, T.; Cherny, R. A.; Bush, A. I. *Exp. Gerontol.* **2000**, *35*, 445.
- (16) Markesbury, W. R.; Carney, J. M. *Brain Pathol.* **1999**, *9*, 133.
- (17) Syme, C. D.; Nadal, R. C.; Rigby, S. E. J.; Viles, J. H. *J. Biol. Chem.* **2004**, *279*, 18169.

transition-metal-catalyzed chemistry associated with AD is thus being actively investigated.

Recent studies show that AD patients exhibit a remarkable 250% increase in heme *b* in their temporal brain.²⁰ It has been proposed that such abnormal heme concentration in the brain of AD patients is because heme, biosynthesized in brain cells, binds to A β , forming heme–A β complexes. This complexation depletes heme, required for biological processes, leading to heme deficiency in cells.²¹ This ultimately results in symptoms such as increases in iron uptake, bilirubin, and heme oxygenase concentration,^{22,23} abnormal iron homeostasis, decay of iron regulatory proteins, dysfunction in mitochondrial complex IV, oxidative stress, etc.^{20,24} These symptoms are all characteristic pathological features of AD. Further it has been shown that the human A β (1–42)–heme complex shows peroxidase activity and can catalyze the oxidation of neurotransmitters such as serotonin, 3,4-dihydroxyphenylalanine, and 4-hydroxyphenylpyruvic acid by H₂O₂. This could be the probable reason for the abnormal neurotransmission observed in AD patients.^{25,26}

The involvement of heme in AD pathology opens up a new dimension in AD research and may hold a key towards development of a cure. However, so far no information is available about the nature and properties of these heme–A β active sites. In this study we use absorption and electron paramagnetic resonance (EPR) spectroscopic techniques to identify the amino acid residues of the human A β peptide that bind to heme, using native human A β (1–40) peptide, isolated peptide fragments A β (1–16), A β (17–40), and A β (10–20), and various site-directed mutants. Peroxidase activity assays provide insight into the nature of the active site environment of these peptide–heme complexes, including the key second-sphere residues involved. We evaluate the effect of pH on these complexes to elucidate the nature of the exchangeable sixth ligand of the heme-bound A β peptide complexes.

Experimental Methods

Materials. All reagents were of the highest grade commercially available and were used without further purification. A β peptides of different chain lengths (1–40, 1–16, 17–40, and 10–20) have been used for this study. All site-directed mutants are of the A β (1–16) peptide. The mutants used were His⁶Gly, His¹³Ala, His¹⁴Ala, Tyr¹⁰Gly, Arg⁵Asn, double mutant His¹³Gly, His¹⁴Gly, and triple mutant His⁶Gly, His¹³Gly, His¹⁴Gly. All peptides were purchased from Syn Bio Science Corp and GL Biochem (Shanghai) Ltd. with >95% purity. Hemin and the buffers were purchased from Sigma. Different buffers were used at different pH ranges: MES for pH 5–6.4, HEPES and phosphate for pH 7–8, and glycine for pH 9–10. All the peptide stock solutions were made in 100 mM phosphate buffer. A β (1–40), A β (17–40), and A β (10–20) peptide stock was made in 20 mM Hepes buffer, while hemin solution was made in 1 M NaOH solution. Peptide stock solutions were 0.5 mM,

and the heme stock solution was 5 mM. Heme–A β complexes were prepared by incubating 1 equiv of both A β and heme solutions for ~1 h. Absorption spectra were recorded by adding 20 μ L of the heme–A β complex solution in 1 mL of buffer. All the spectral data were obtained by a UV–vis diode array spectrophotometer (Agilent 8453).

EPR spectra were obtained by a Jeol (JES FA200) spectrophotometer. EPR samples were 0.5 mM in concentration and were run at 77 K in a liquid nitrogen finger dewar. Low-pH EPR samples were prepared by lowering the pH of a high-pH sample with 0.5 M H₂SO₄.

Peroxidase Activity Measurement. 3,3',5,5'-Tetramethylbenzidine (TMB) was used as the substrate for peroxidase activity measurement. A 10 mg portion of TMB was dissolved in 0.5 mL of glacial AcOH and 10 mL of AcOH/NaOAc buffer (1 M, pH 4.5). The solution was diluted to 25 mL with water. This was followed by addition of 100 μ L of 30 volume H₂O₂. A 10 μ L volume of 0.05 mM protein sample was added to the above solution. Kinetic traces were obtained by monitoring the increase of the 652 nm absorption band with time.²⁷

Catalase Activity Measurement. Catalase activity was measured by following the rate of decay of H₂O₂ at 240 nm. A 1.5 mL volume of deionized water and 100 μ L of 0.05 M H₂O₂ were mixed in a cuvette. A 20 μ L volume of 0.5 mM protein was added to it. All spectra were recorded in the kinetic mode, with continuous stirring at 25 °C.²⁸

Results and Analysis

Choice of Peptides. Human A β (1–40) was the chosen parent peptide for this study as it is the most common A β peptide found physiologically. Two fragments, the hydrophilic A β (1–16) and the lipophilic A β (17–40), have been separately considered to narrow the heme binding region (Scheme 1A). There are four residues in A β (1–40) that are known to bind heme biologically. They are the three histidine residues (His binds heme in hemoglobin, myoglobin, cytochrome *c* oxidase, etc.) His⁶, His¹³, and His¹⁴ and a tyrosine residue (Tyr binds heme in catalase), Tyr¹⁰, which are all present in the A β (1–16) fragment (Scheme 1C). To identify the binding residue, site-directed mutants of A β (1–16) have been selected (Scheme 1D). Single mutants His⁶Gly (i.e., Tyr¹⁰, His¹³, and His¹⁴ are the available coordinating residues), His¹³Ala (i.e., His⁶, Tyr¹⁰, and His¹⁴ are the available coordinating residues), His¹⁴Ala (i.e., His⁶, Tyr¹⁰, and His¹³ are the available coordinating residues),²⁷ and Tyr¹⁰Gly (i.e., His⁶, His¹³, and His¹⁴ are the available coordinating residues) have been used to observe the effect of elimination of the possible binding ligands individually. Double mutant His¹³Gly, His¹⁴Gly (i.e., His⁶ and Tyr¹⁰ are the available coordinating residues) shows the effect of His⁶ and Tyr¹⁰ residues, and triple mutant His⁶Gly, His¹³Gly, His¹⁴Gly (i.e., Tyr¹⁰ is the available coordinating residue) has been chosen to isolate the effect of the Tyr¹⁰ residue (Scheme 1D).

Absorption Spectroscopy. 1 equiv of heme covalently binds with 1 equiv of A β peptide.²⁶ The absorption spectra of 1 equiv of heme-incubated A β (1–40) peptide and the isolated peptide fragments A β (1–16) and A β (17–40) overlaid with heme are presented in Figure 1. The intensity decrease of the Soret band at ~365 nm, the blue shift of the Q-band from ~618 to ~606 nm, and development of a new band at ~693 nm relative to native heme indicate the formation of a heme–A β complex for A β (1–40) (Figure 1, red) and A β (1–16) (Figure 1, green). However, heme does not bind to A β (17–40) as indicated by

(18) Karr, J. W.; Akintoye, H.; Kaupp, L. J.; Szalai, V. A. *Biochemistry* **2005**, *44*, 5478.

(19) Karr, J. W.; Szalai, V. A. *J. Am. Chem. Soc.* **2007**, *129*, 3796.

(20) Atamna, H.; Frey II, W. H. *Proc. Natl. Acad. Sci. U.S.A.* **2004**, *101*, 11153.

(21) Atamna, H.; Liu, J.; Ames, B. N. *J. Biol. Chem.* **2001**, *276*, 48410.

(22) Kimpara, T.; Takeda, A.; Yamguchi, T.; Arai, H.; Okita, N.; Takase, S.; Sasaki, H.; Itoyama, Y. *Neurobiol. Aging* **2000**, *21*, 551.

(23) Schipper, H. M.; Cisse, S.; Stopa, E. G. *Ann. Neurol.* **1995**, *37*, 758.

(24) Steffens, G. C.; Biewald, R.; Buse, G. *Eur. J. Biochem.* **1987**, *164*, 295.

(25) Atamna, H.; Boyle, K. *Proc. Natl. Acad. Sci. U.S.A.* **2006**, *103*, 3381.

(26) Atamna, H.; Frey, W. H., II; Ko, N. *Arch. Biochem. Biophys.* **2009**, *487*, 59.

(27) Holland, V. R.; Saunders, B. C.; Rose, F. L.; Walpole, A. L. *Tetrahedron* **1974**, *30*, 3299.

(28) Beers, R. F., Jr.; Sizer, I. W. *Science* **1954**, *120*, 32.

Scheme 1. Amino Acid Sequence of Human (A) A β (1–40) Showing the Hydrophilic Region in Blue and the Lipophilic Region in Yellow, (B) A β (10–20), (C) A β (1–16) Highlighting the Residues That Have Been Mutated in This Study, and (D) Mutants of the A β (1–16) Peptide



the absence of spectral changes relative to heme (Figure 1, blue) even after prolonged incubation. This is consistent with the absence of a heme-binding residue in this lipophilic peptide sequence. The similar nature of the spectra for both A β (1–40) and A β (1–16) indicates that the heme-binding residue lies within the 1–16 peptide sequence. A β (10–20) (Scheme 1B) binds heme and results in a spectrum similar to that of the A β (1–40) and A β (1–16) peptides (Figure 1, yellow). This indicates that the coordinating residue lies between Tyr¹⁰ and Lys¹⁶.

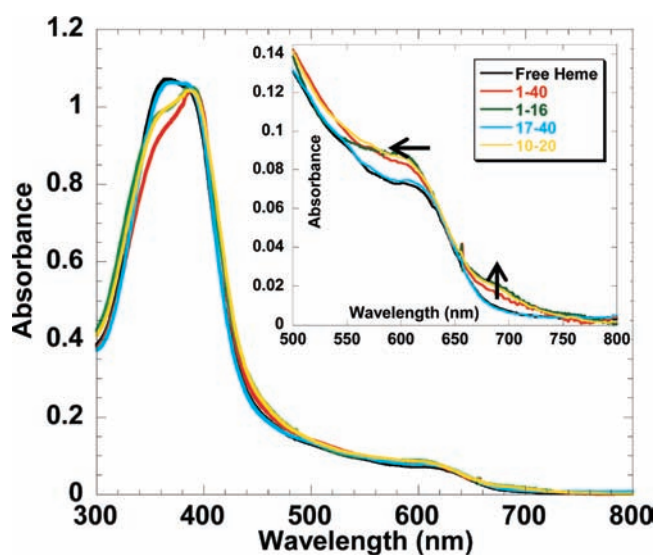


Figure 1. Absorption spectra of heme-incubated A β peptides at pH 7: (1–40), red; (1–16), green; (17–40), blue; (10–20), yellow; free heme, black. Inset: enlarged Q-band region showing changes as indicated by arrows relative to heme.

One molecule of A β (1–40) and A β (1–16) quantitatively binds one molecule of heme. The triple mutant His⁶Gly, His¹³Gly, His¹⁴Gly when incubated with heme shows no spectral changes relative to heme, indicating that it does not bind heme (Figure 2, green). The fact that Tyr¹⁰ is the only available coordinating residue in this mutant peptide clearly indicates that Tyr¹⁰ is not a coordinating ligand for these A β peptides. This has been further probed by the Tyr¹⁰Gly mutant, which binds heme like native A β (1–40), as reflected

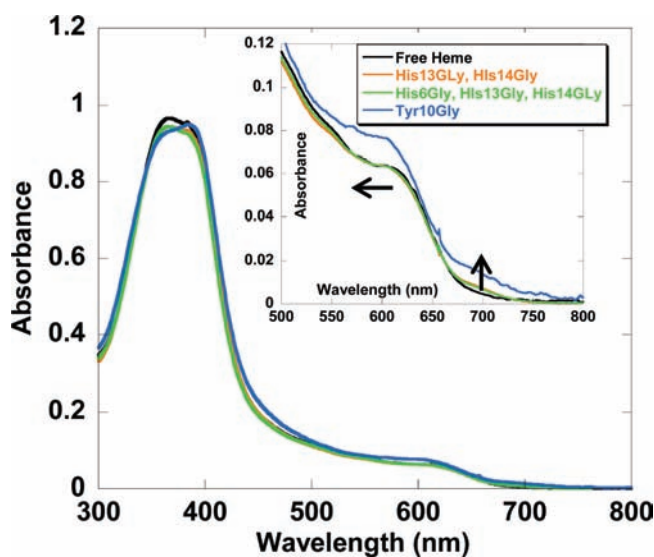


Figure 2. Absorption spectra of heme-incubated mutants of A β (1–16) peptide at pH 7: double mutant His¹³Gly, His¹⁴Gly, orange; triple mutant His⁶Gly, His¹³Gly, His¹⁴Gly, green; Tyr¹⁰Gly, blue; free heme, black. Inset: enlarged Q-band region showing changes as indicated by arrows relative to heme.

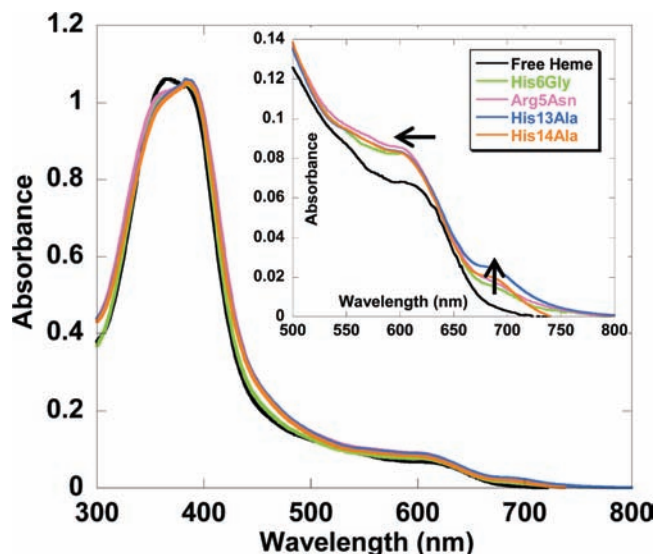


Figure 3. Absorption spectra of heme-incubated mutants of $A\beta(1-16)$ peptide at pH 7: His⁶Gly, green; Arg⁵Asn, pink; His¹³Ala, blue; His¹⁴Ala, orange; free heme, black. Inset: enlarged Q-band region showing changes as indicated by arrows relative to heme.

from its absorption spectrum (Figure 2, blue), confirming that Tyr¹⁰ is not the residue binding heme. Thus, one of the three available histidine residues (His⁶, His¹³ or His¹⁴) coordinates heme. The double mutant His¹³Gly, His¹⁴Gly does not bind heme (Figure 2, orange), similar to the triple mutant. Thus His⁶, available for binding heme, is not the coordinating residue either. This means that either His¹³ or His¹⁴ is the heme-binding ligand. This has also been established by the single residue mutant His⁶Gly (with Tyr¹⁰, His¹³, and His¹⁴ as available coordinating residues) (Figure 3, green), which binds heme, implying that either His¹³ or His¹⁴ binds heme. To distinguish between His¹³ and His¹⁴ residues, the single mutants His¹³Ala and His¹⁴Ala were selected. The absorption spectra of both the above heme- $A\beta$ mutants show spectral features characteristic of heme-bound forms (Figure 3, blue and orange). Thus, either His¹³ or His¹⁴ is capable of binding heme. Absorption spectroscopy is not sensitive enough to distinguish between His¹³ and His¹⁴ coordination.

EPR Spectroscopy. EPR spectroscopy is a powerful technique that is sensitive to small changes in the electronic structure of paramagnetic species. Figure 4 shows the 77 K, X-band EPR spectra of the heme- $A\beta(1-16)$ complex overlaid with free heme. The EPR spectrum of the heme-bound peptide is shifted to lower field relative to that of heme, indicating the formation of heme-bound species. The spectrum of the heme-peptide complex shows an axial high-spin ($S = 5/2$) Fe signal with $g \approx 6.0$. No high-field, low-spin signals were observed at this temperature. These data also show the presence of rhombic splitting in the primarily axial $g \approx 6.0$ signal.²⁹ Thus, the $A\beta$ peptide binds heme through histidine ligands (either His¹³ or His¹⁴) and forms an axial high-spin Fe^{III} active site with rhombic distortion like those found in hemoglobin, myoglobin, and several peroxidases.³⁰⁻³²

EPR measurements were performed on the heme- $A\beta$ complexes of the His¹³Ala (i.e., heme bound to His¹⁴) and His¹⁴Ala

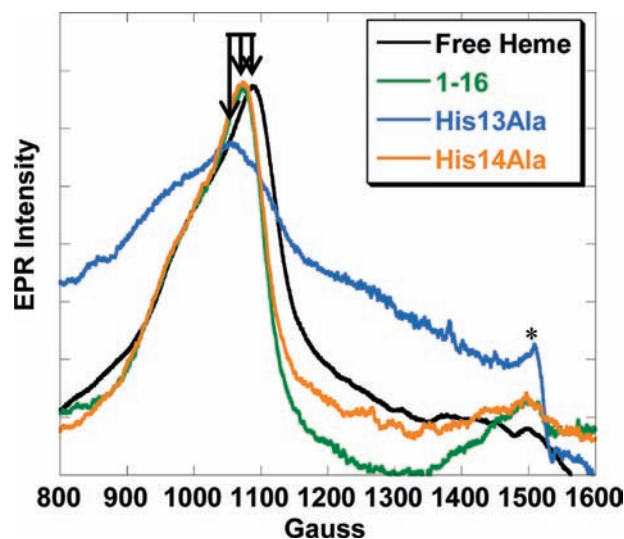


Figure 4. EPR spectra of heme- $A\beta$ complexes at 77 K: (1-16), green; His¹³Ala mutant of (1-16), blue; His¹⁴Ala mutant of (1-16), orange; free heme, black. Arrows indicate the shift in the EPR signal. The asterisk indicates a negligible amount of rhombic high-spin signal observed at $g \approx 4.3$.

(i.e., heme bound to His¹³) mutants of $A\beta(1-16)$ to spectroscopically differentiate between the two possible histidine ligands (His¹³ and His¹⁴) that have been identified as the heme-binding ligands from absorption experiments (vide supra). The EPR spectrum of the His¹⁴Ala mutant of $A\beta(1-16)$ peptide shows similar features compared to that of the $A\beta(1-16)$ peptide (Figure 4, orange and green, respectively). Thus, the His¹³ residue binds heme in the His¹⁴Ala mutant, resulting in an active site which is spectroscopically similar to that of native heme- $A\beta(1-16)$. However, the EPR spectrum of the His¹³Ala mutant (Figure 4, blue) is different from those of the native heme- $A\beta$ complexes and the His¹⁴Ala mutant where the spectrum is more rhombic (i.e., the g_1 and g_2 components of $g_{\text{perp}} \approx 6.0$ signal are split more) relative to that of heme- $A\beta(1-16)$ (Figure 4, green) and less intense.³³ The difference between the EPR data of His¹³Ala and those of His¹⁴Ala and native $A\beta(1-16)$ -heme complexes indicates that His¹⁴ binds heme when His¹³ is unavailable. Thus, we conclude that, though His¹³ and His¹⁴ can bind heme, when both residues are present as in $A\beta(1-16)$, His¹³ probably binds heme preferentially.

Peroxidase Activity. The peroxidase (histidine-coordinated heme enzyme) activities of the native heme- $A\beta$ complexes and their mutants and free heme were tested by following the catalytic oxidation of the substrate TMB by H₂O₂. The heme complexes of $A\beta(1-40)$ and $A\beta(1-16)$ show similar peroxidase activity (Figure 5, red and green, respectively), further indicating similar active site environments of the heme-bound complexes in these peptides. This peroxidase activity is similar to reported values of heme complexes of

(30) Rotilio, G.; Calabrese, L.; Giacometti, G. M.; Brunori, M. *Biochim. Biophys. Acta* **1971**, *236*, 234.

(31) Peisach, J.; Blumberg, W. E.; Wittenberg, B. A.; Wittenberg, J. B.; Kampa, L. *Proc. Natl. Acad. Sci. U.S.A.* **1969**, *63*, 934.

(32) Blumberg, W. E.; Peisach, J.; Wittenberg, B. A.; Wittenberg, J. B. *J. Biol. Chem.* **1968**, *243*, 1854.

(33) The reduced signal intensity of the His¹³ mutant could result from its increased rhombicity or small changes in the axial ZFS tensor, D , which could lead to greater population of the $g_{\text{perp}} = 6$ EPR excited state.

(29) Note that due to the low resolution of the data at 77 K we cannot accurately determine the E/D parameter. However, the data show changes that can be used for diagnostic purposes.

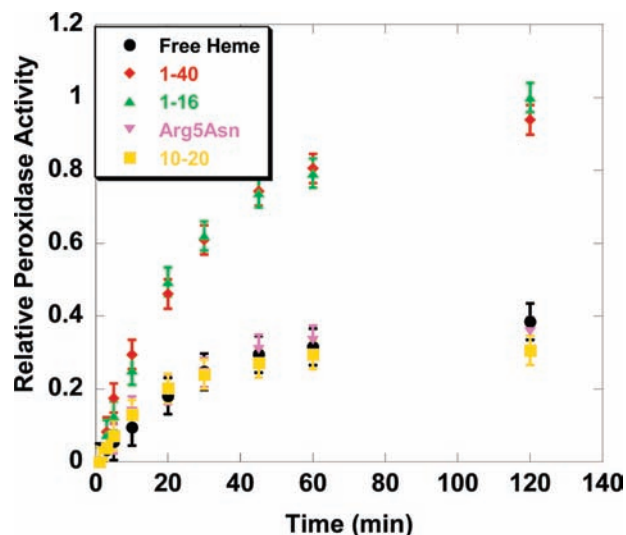


Figure 5. Kinetic traces for peroxidase activity, monitoring the increase of the 652 nm absorbance intensity, for different heme–A β complexes: (1–40), red; (1–16), green; (10–20), yellow; Arg⁵Asn mutant of (1–16), pink; free heme, black.

A β (1–42) and is ~ 3 times faster compared to that of free heme.²⁵ The triple mutant His⁶Gly, His¹³Gly, His¹⁴Gly and double mutant His¹³Gly, His¹⁴Gly show peroxidase activity comparable to that of free heme (Figure S1A, Supporting Information). This is consistent with the fact that these mutants do not bind heme. The single mutants of the A β (1–16) peptide, His⁶Gly, His¹³Ala, and His¹⁴Ala, bind heme and show activity comparable to that of the wild-type A β (1–16) and A β (1–40) peptides (Figure S1B). This implies that, though the active sites are not identical for His¹³- and His¹⁴-bound heme complexes, the peroxidase activity is not affected significantly by changing the coordinating histidine residues.

Interestingly, when peroxidase activity was tested for the heme–A β (10–20) complex (Figure 5, yellow), it did not show any enhanced activity compared to heme (Figure 5, black), in contrast to the other heme–A β complexes of peptides that bind heme. This is counterintuitive, because the A β (10–20) peptide contains both the coordinating histidine residues (His¹³, His¹⁴). This suggests that one or more noncoordinating residues in the 1–9 peptide region are essential for the heme–A β complexes to function as a peroxidase. It is well-known that peroxidase activity requires the presence of an acidic arginine residue in the distal side of heme which is protonated at physiological pH and provides the proton required for the heterolytic cleavage of the O–O bond of H₂O₂.^{34,35} It is possible that the acidic Arg⁵ residue might be in the distal pocket of heme and serve as the proton source required for peroxidase activity.

In fact, it was observed that the Arg⁵Asn mutant of A β (1–16) binds heme (Figure 3, pink); however, the heme-bound complex when tested for peroxidase activity showed no enhanced activity compared to native heme, comparable to A β (10–20) (Figure 5, pink). This clearly demonstrated that Arg⁵ is the key distal residue required for these heme–A β complexes to function as peroxidases.

Catalase Activity. The catalase (tyrosine-coordinated heme enzyme) activities of the heme–A β complexes have been tested, and it has been found that these heme complexes do not show any catalase activity with respect to free heme (Figure S2, Supporting Information). This further supports the conclusion drawn from spectroscopy that Tyr¹⁰ does not coordinate to heme to form catalase-type active sites.³⁶

pH Perturbation of the Trans Axial Ligand. The observed peroxidase activity, which requires binding of H₂O₂ to the heme Fe trans to the axial His¹³ ligand, implies the presence of an exchangeable sixth ligand, trans to the axial histidine. The nature of this sixth ligand was investigated by pH perturbation.

The heme–A β (1–40) and heme–A β (1–16) complexes exhibit decreases in the Soret band (~ 370 nm) and Q-band (~ 635 nm) intensities and a blue shift of the Q-band on going from low to high pH with a pK_a of $\sim 6.8 \pm 0.3$ (Figures 6A,B and 7). The pH effect of heme–A β (10–20) shows similar spectral changes (Figure 6C), but the pK_a obtained was $\sim 8.2 \pm 0.3$ (Figure 7), which is similar to that observed for myoglobin (Figure S3, Supporting Information). This observed pH effect (pK_a ≈ 8.2) likely reflects the protonation equilibrium of a water-derived ligand (as in the case of myoglobin) at the distal side of the heme; i.e., there is a water bound at low pH and a hydroxide bound at high pH (Figure S4, Supporting Information).³⁷ It is worth mentioning that the pK_a of the metal-bound histidine ligand (His¹³) is much lower than the observed pK_a values.³⁸ The higher pK_a observed for A β (10–20) is likely due to the absence of the Arg⁵ residue. In fact, the Arg⁵ mutant exhibits a pK_a of $\sim 8.2 \pm 0.2$ (Figures 6D and 7), similar to that of A β (10–20). This is consistent with the presence of H-bonding interaction between the water-derived trans axial ligand and the distal Arg⁵ residue. This H-bonding with the positively charged protonated guanidine side group of Arg⁵ will stabilize a bound hydroxide more than the water, thus lowering its pK_a. A similar lowering of the pK_a of a heme-bound axial water ligand by H-bonding to an arginine residue has also been observed in the Thr⁶⁷Arg mutant of myoglobin.³⁹

Discussion

The results presented in this study indicate that a residue in the hydrophilic 1–16 region of human A β peptide binds heme (Scheme 1, Figure 1). EPR data indicate that this forms a high-spin Fe^{III} active site (Figure 4; Figure S5, Supporting Information). In the absence of all the possible coordinating histidine residues (triple mutant His⁶Gly, His¹³Gly, His¹⁴Gly), heme does not bind to the A β peptides, implying that one of the three histidine residues (His⁶, His¹³, or His¹⁴) binds heme (Figure 2). This further implies that the only available coordinating residue, Tyr¹⁰, does not bind heme, which is also supported by the fact that a tyrosine-depleted mutant (Tyr¹⁰Gly) binds heme (Figure 2). Also these complexes do not show any catalase activity (Figure S3, Supporting Information), which requires the coordination of the tyrosine residue. Mutating the His¹³ and His¹⁴ residues simultaneously

(34) Rodríguez-López, J. N.; Smith, A. T.; Thornley, R. N. *F. J. Biol. Chem.* **1996**, *271*, 4023.

(35) Henriksen, A.; Smith, A. T.; Gajhede, M. *J. Biol. Chem.* **1999**, *274*, 35005.

(36) Murthy, M. R. N.; Reid, T. J., III; Sicignano, A.; Tanaka, N.; Rossmann, M. G. *J. Mol. Biol.* **1981**, *152*, 465.

(37) Note that EPR data on the high- and low-pH heme–A β complexes show increased rhombicity in the low-pH form.

(38) Teale, F. W. J. *Biochim. Biophys. Acta* **1959**, *35*, 543.

(39) Redaelli, C.; Monzani, E.; Santagostini, L.; Casella, L.; Sanangalantoni, A. M.; Pierattelli, R.; Banci, L. *ChemBioChem* **2002**, *3*, 226.

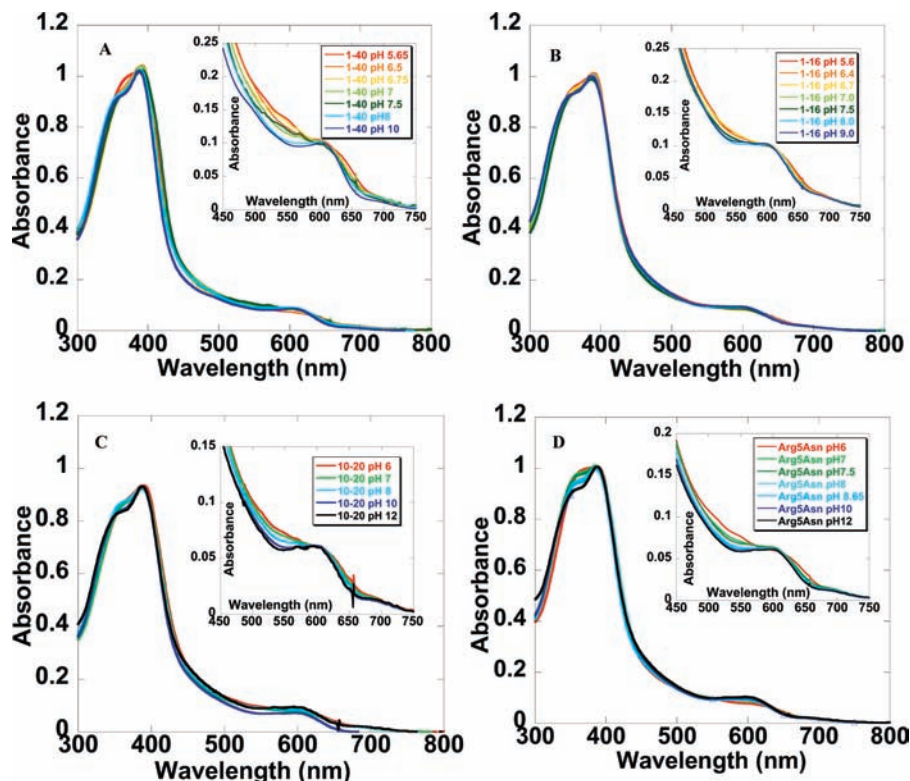


Figure 6. Absorption spectra of heme- $A\beta$ complexes at different pH values: (A) $A\beta(1-40)$, (B) $A\beta(1-16)$, (C) $A\beta(10-20)$, (D) Arg⁵Asn mutant of $A\beta(1-16)$. Inset: enlarged Q-band region.

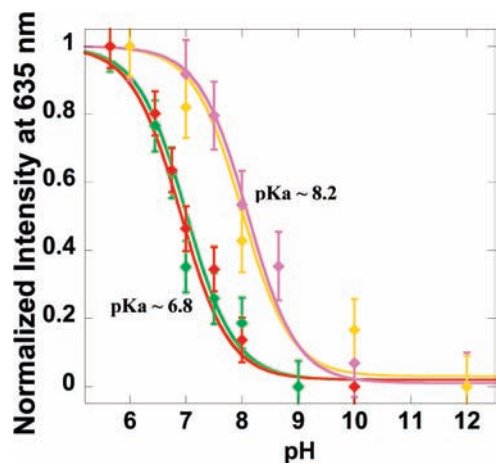


Figure 7. pK_a plots of heme- $A\beta$ complexes with error bars: (1-40), red; (1-16), green; (10-20), yellow; Arg⁵Asn mutant of (1-16) peptide, pink. All data points from Figure 6 have been used for each pK_a determination.

(double mutant His¹³Gly, His¹⁴Gly) prohibits heme binding, showing that His⁶ does not bind heme (Figure 2). The His¹³Ala mutant binds heme, and so does His¹⁴Ala, as determined from absorption spectroscopy (Figure 3). This implies that either His¹³ or His¹⁴ can coordinate to heme. EPR spectra of the heme complexes of these two mutants (Figure 4) show that the His¹⁴Ala (His¹³ being the only available coordinating residue) mutant exhibits an EPR spectrum very similar to that of the heme complex of the native $A\beta(1-16)$, but the His¹³Ala (His¹⁴ being the only other coordinating residue) mutant exhibits higher rhombicity than the heme complex of the native $A\beta(1-16)$ pep-

Scheme 2. Amino Acid Sequences of Human, Pig, Monkey, Rabbit, etc. (Top) and Rat, Mouse, and Chinese Hamster (Bottom) $A\beta(1-16)$ Highlighting the Amino Acids that Differ between the Two Peptide Sequences

Asp-Ala-Glu-Phe-Arg⁵-His-Asp-Ser-Gly-Tyr¹⁰-Glu-Val-His¹³-His-Gln-Lys
Asp-Ala-Glu-Phe-Gly⁵-His-Asp-Ser-Gly-Phe¹⁰-Glu-Val-Arg¹³-His-Gln-Lys

ptide. The increased rhombicity can arise from differences in the orientation of the imidazolyl ring of these two histidine residues with respect to the g tensors or differences in the second-sphere environment of the active site.⁴⁰ While the exact origin of this effect is unclear, the data indicate that probably His¹³ binds heme in the native $A\beta(1-16)$ peptide.

The sixth ligand of the heme-bound $A\beta$ complexes has an exchangeable water-derived ligand, which has a pK_a of ~ 6.8 for both $A\beta(1-40)$ and $A\beta(1-16)$ (Figures 6 and 7). However, $A\beta(10-20)$ and the Arg⁵Asn mutant of $A\beta(1-16)$, both of which lack the Arg⁵ residue, show similar spectral changes on perturbing the pH, but with a higher pK_a (~ 8.2). This lowering of pK_a in the presence of Arg⁵ is consistent with the presence of H-bonding interaction between the heme-bound water ligand and the arginine residue, which stabilizes the hydroxide form more than the water form.

The peroxidase activities of the $A\beta(1-40)$ -heme and $A\beta(1-16)$ -heme complexes show enhanced activity relative to free heme, suggesting the formation of a histidine-bound peroxidase-like active site (Figure 5). However, $A\beta(10-20)$, though it binds heme, fails to show any enhanced peroxidase activity unlike the other $A\beta$ peptides. Peroxidases have a conserved arginine residue in the distal pocket of the active site,

(40) Ikeda-Saito, M.; Horill, H.; Andersson, L. A.; Prince, R. C.; Pickering, I. J.; George, G. H.; Sanders, C. R.; Lutz, R. S.; McKelvey, E. J.; Matterna, R. *J. Biol. Chem.* **1992**, *267*, 22843.

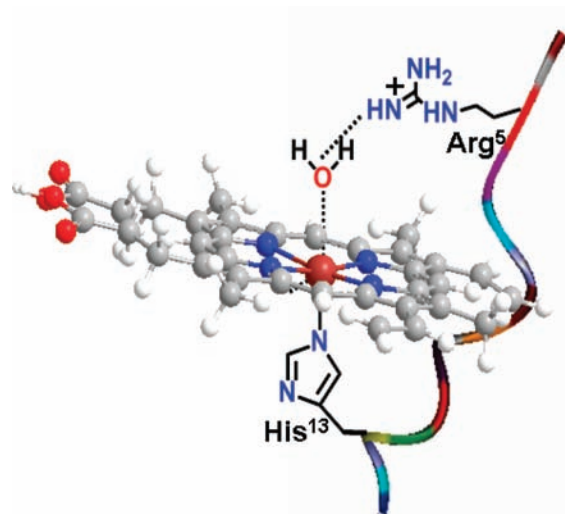


Figure 8. Schematic representation of the active site environment of heme-A β (1-16), which may be extrapolated for native A β (1-40).

which provides a proton for the O–O bond cleavage in the catalytic cycle. A β (10–20) lacks an arginine residue (Arg⁵) which is present in both A β (1–40) and A β (1–16) and might be crucial for peroxidase activity. In fact, an Arg⁵Asn mutant binds heme, but shows no peroxidase activity, confirming its role in making these heme–A β complexes function as peroxidases. It is important to note that rodent (i.e., rats, mouse, etc.) A β , which lacks this Arg⁵ residue (Scheme 2), binds heme but shows no marked enhancement in peroxidase activity relative to free heme,²⁶ similar to A β (10–20) and Arg⁵Asn peptides, consistent with the above conclusion. It is also important to note that His¹³ is absent in rodent A β ; thus, His¹⁴ could be the probable coordinating ligand. The fact that the two key residues (Arg⁵ and His¹³) involved in the heme–A β active site of humans/pigs/monkeys are absent in rodent A β peptides is very

interesting, as AD does not affect these rodents.⁴¹ This further stresses the importance of heme binding to A β in the pathology of AD.

Summary

In summary, we have established that either His¹³ or His¹⁴ of A β peptide can bind heme at the proximal side. However, in the native human A β (1–16) peptides, His¹³ probably binds heme specifically (Figure 8). The open distal position of the heme–A β complex has an exchangeable water-derived ligand with a p*K*_a of 6.8. The acidic Arg⁵ residue present at the distal side directly H-bonds with the axial water ligand, lowering its p*K*_a. It also induces a pull effect by providing a proton required to drive the O–O bond heterolysis, hence making these heme–A β complexes function as peroxidases. These findings shed light on the nature of interaction of heme with A β peptides, which is a key step toward unraveling the role played by heme in AD. The two A β residues critical for the active site of the heme–A β complex, His¹³ and Arg⁵, are both absent in the A β of rodents, which do not show AD, further illustrating that heme may play a greater role in AD than previously thought.

Acknowledgment. We thank the Indian Association for the Cultivation of Science for funding this study through a start-up grant. D.P. is thankful to the Council of Scientific and Industrial Research, India, for a Junior Research Fellowship.

Supporting Information Available: Peroxidase activity of all the peptides and mutants, catalase activity kinetic traces and absorption spectra, pH effect (absorption and EPR spectra) and p*K*_a plot of myoglobin, EPR spectra of A β (1–16) at low and high pH, EPR spectra of the heme complex of the Arg⁵Asn mutant of A β (1–16), extended absorption spectra of all heme–A β complexes, and high-field EPR spectra of heme–A β complexes. This material is available free of charge via the Internet at <http://pubs.acs.org>.

JA1084578

(41) Vaughan, D. W.; Peters, A. *J. Neuropathol. Exp. Neurol.* **1981**, *40*, 472.